

Overview of EmeraChem:

Progress on
NOx Absorber Sulfur Tolerance
and ORNL Activities to Advance Development

Jim Parks - EmeraChem

Bill Partridge, Todd Toops, Sam Lewis,

John Storey, Karren More – Oak Ridge National Laboratory

Neal Currier, Jason Chen, Roger England – Cummins, Inc.

Achieving Diesel Product Status with NOx Absorber

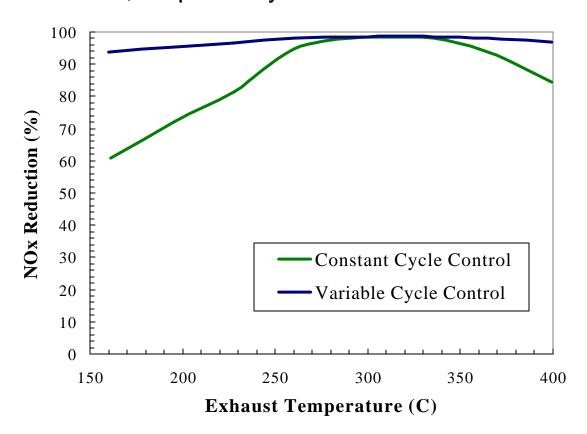
- NOx Absorber Technology is Clean Enough for NOx Emission Regulations
 - <0.2 g/bhp-hr NOx Emissions Demonstrated on 6.9 g/bhp-hr NOx Engine
- Durability is Main Criteria to be Demonstrated
 - Heavy-Duty Regulation Calls for 435,000 Mile Durability
- Most Significant Durability Issue for NOx Absorbers
 - Sulfur Masking of NOx Storage Sites

Key Attributes of EmeraChem NOx Absorber

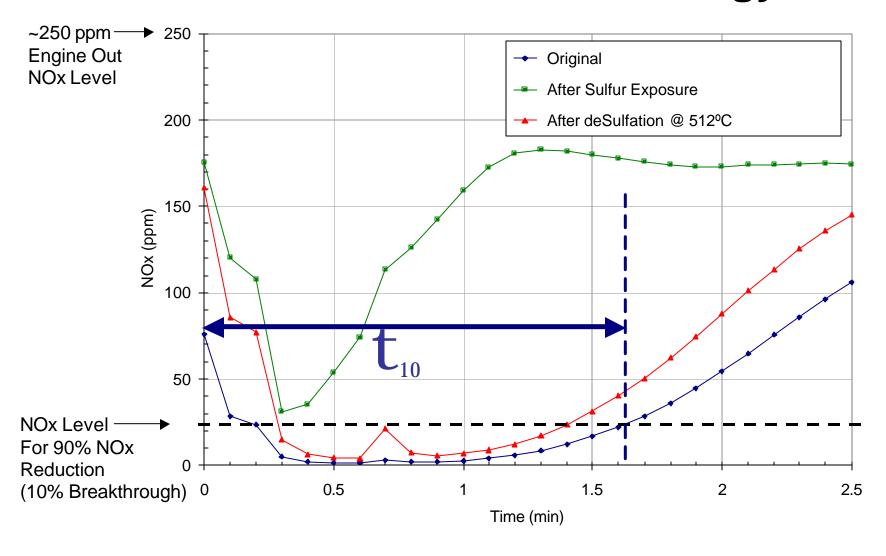
- High NOx Reduction Efficiencies Over a Broad Range of Temperatures Well Suited to Diesel Engine
 - Can Use Diesel Fuel for Catalyst Reductant
- deSulfation at Relatively Low Temperatures (<550°C)
 - Minimizes Thermal Degradation Occurring During deSulfation
 - Lowers Fuel Penalty for deSulfation
 - Reduces Complications Associated with deSulfation

NOx Absorber Performance for Diesel Exhaust

- Data Obtained on 11.9 g/bhp-hr NOx Engine with Diesel Fuel
 Reductant
- Constant Cycle Control Data Represents Catalyst Capacity
- Variable Cycle Control Data Represents Engineered Performance
- •Fuel Penalties for Constant Cycle and Variable Cycle Data were Similar
 - •3.6% and 3.8%, respectively

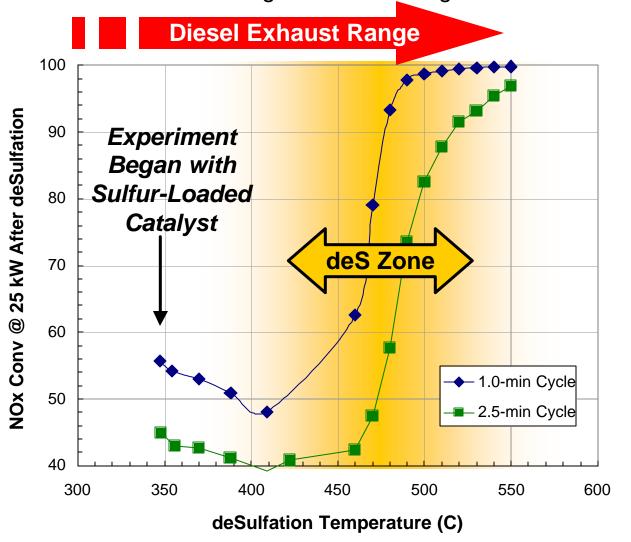


deSulfation Recovery of NOx Capacity: NOx Profiles and Terminology



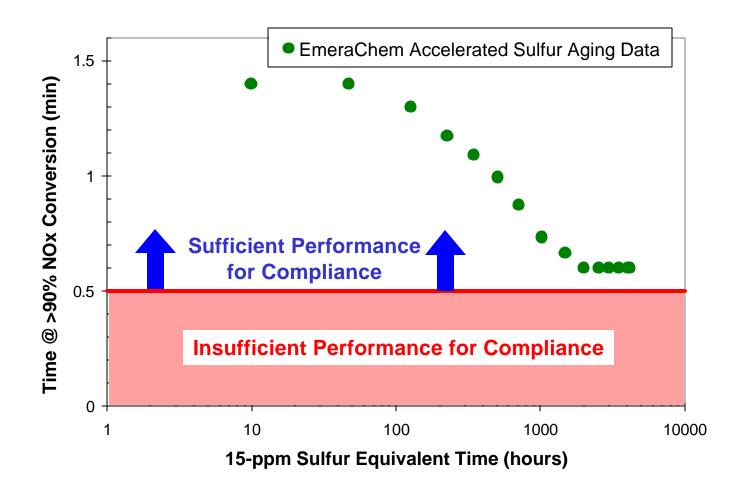
deSulfation in Diesel Exhaust Range

- •Data from 6.9 g/bhp-hr NOx Engine with NOx Absorber Only in System
- •Diesel Fuel Reductant Used for Regeneration During deSulfation



Aging on Engine with 150-ppm S Fuel (DECSE)

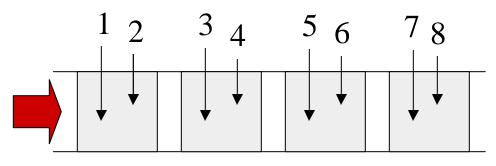
- Ability to Obtain >90% NOx Reduction Efficiency Maintained
 - •Performance Stabilizes After Initial Thermal Degradation Stabilizes



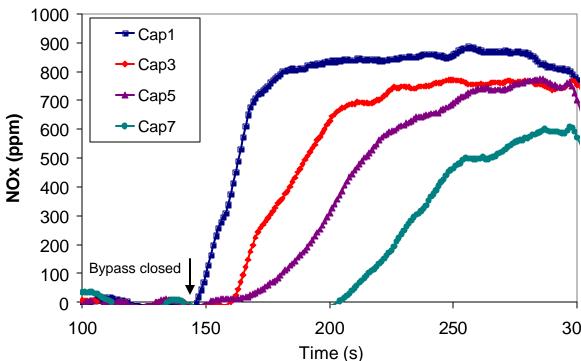
Path to NOx Absorber Product

- While Progress Has Been Made in Sulfur Tolerance for the NOx Absorber Catalyst...
- Development of a NOx Absorber Catalyst Product Will Require:
 - More Understanding of Fundamental Catalyst Mechanisms
 - Characterization of ALL Aging Mechanisms
- US DOE Labs Offer Unique Capabilities to Assist in Development Path

SpaciMS Exhaust Analysis with NTRC



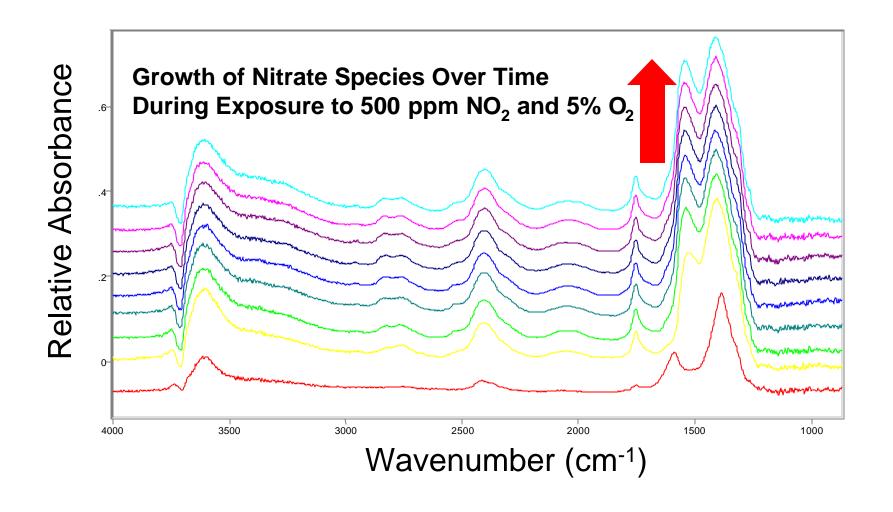
Exhaust Sampled by8 Capillaries DistributedAlong Flow Axis



- •Front third of first catalyst saturates within ca. 20 sec.
- Last catalyst is not used in initial 60 sec.
- Profiles can be used to verify and develop
 300capacity-loading models

DRIFT - Surface Chemistry with NTRC

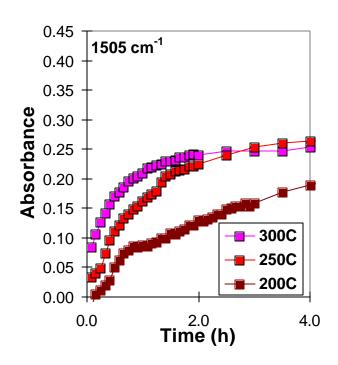
- •DRIFT Detects Unique Solid State NOx Absorber Surface Chemistry
- •Technique Targeted for Characterization of Nitrate/Carbonate/Sulfates



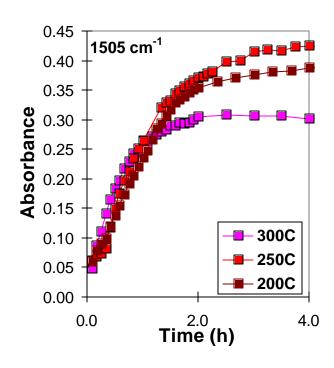
Interesting Differences Observed by DRIFT

- •Differences Observed in Surface Chemistry for NO vs. NO₂ Exposure
- •More Data Needed to Fully Understand NOx Absorber Mechanism
- •Fundamentals Will Be Useful in Optimizing Catalyst Performance and Cost

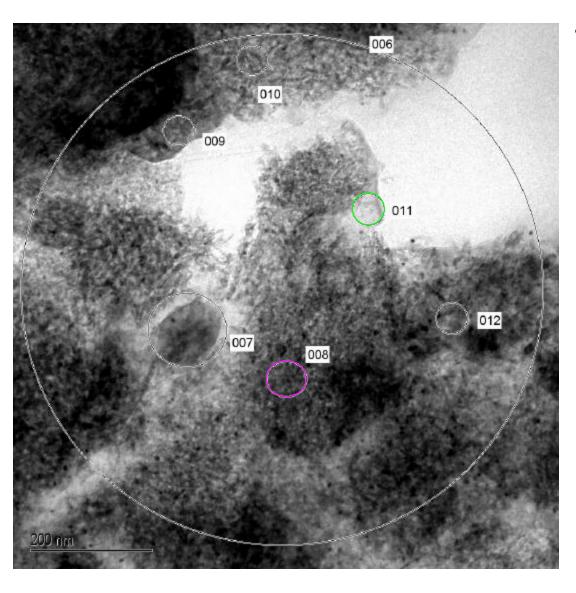
$$500 \text{ ppm NO} + 5\% \text{ O}_{2}$$



$$500 \text{ ppm NO}_2 + 5\% \text{ O}_2$$



Catalyst Morphology Characterization with HTML



- TEM Image of K-based NOx Absorber Catalyst
 - Catalyst Was Not Aged
- Precious Metals FinelyDispersed Over Washcoat
- Majority of K Sorbate Finely Dispersed
- •Some Amorphous K Deposits Occur at Interfaces
- •Future Studies Will Characterize Aged Catalysts

K Concentration via X-Ray Analysis (a.u.):

Position 011: 560 Position 008: 120

Noise Level: 10

Summary

- NOx Absorber Technology Obtains High NOx Reduction Efficiencies
- Control of Sulfur Effects Demonstrated in Accelerated Aging Studies
- A Greater Understanding of Catalyst Fundamentals Will Lead to an Advanced Cost-Effective Product